Pot marigold flower extract (*Calendula officinalis* L.) has pharmacological properties due to the presence of various bioactive compounds. It is known that the extract has antioxidant, anti-inflammatory, antitumor, antibacterial, antifungal, antiviral, antimutagenic, antidermatitis properties, etc. The aim of this study was to improve the quality of the selected topical formulation by adding the ethanolic extract of pot marigold flower, as well as to monitor its stability. The topical formulation was water-in-oil emulsion prepared using the hot/hot emulsification process with an oil phase consisting of Vaseline, lanolin, and almond oil. The extract, prepared by ultrasound-assisted extraction, had an antioxidants content of 3.512 g gallic acid equivalent per 100 g dry weight and the half-maximal inhibitory concentration of 0.14 mg mL$^{-1}$ determined by the DPPH assay. Chemical stability studies have shown that daylight has no significant effect on the stability of antioxidants in the extract, while an increase in temperature leads to their degradation. The shelf-life of the extract is about 8 months at 4 °C and 3 months at 22 °C (room temperature). The prepared uncategorized topical formulations containing 1% and 2% (w/w) pot marigold extract were stable at different temperatures during the storage. The uncategorized formulations showed antioxidant activity, but the activity of the extract in the formulations decreased with increasing storage temperature. Pot marigold flower extract and the developed uncategorized formulations showed an inhibitory effect on Gram-positive (*Staphylococcus aureus*) and Gram-negative bacteria (*Escherichia coli*, *Proteus mirabilis*, *Klebsiella pneumoniae*), as well as on *Candida albicans*. The uncategorized formulations with this activity can be used in the treatment of skin infection.

**Keywords:** extract, emulsion, antioxidant activity, antimicrobial activity, chemical stability, quality control.

**Introduction**

Recent studies give favor plant extracts mainly due to the limited use of ingredients of animal origin and the increased demand for organic and sustainable products [1]. Their incorporation into the topical formulations is important since they are a source of vitamins, polyphenols, terpenes, carotenoids, glycosides, saponins, essential oils, and other bioactive compounds. They have beneficial effects on the skin that are usually synergistic [2, 3]. Bioactive compounds that are applied to the skin work by softening the epidermis, improving blood circulation and enabling rapid epithelialization and regeneration. The topical formulations should be emollient, anti-inflammatory, antiseptic, and stimulate skin regeneration. The active ingredients added to the products are mostly herbal extracts (pot marigold, chamomile, *etc.*) and oils (almond, common evening-primrose, *etc.*).

Pot marigold flower extract is commonly used in traditional medicine due to the presence of carotenoids, flavonoids, essential oil [4], pentacyclic triterpenes [5], alkaloids, and tannins [6]. The use of this extract is very popular in the development of pharmaceutical and cosmetic products for topical application to the skin [7]. It helps in the synthesis of collagen and the regeneration of skin cells, which makes it suitable for the treatment of wounds [8], rashes, sunburns, inflammatory processes on the skin, insect bites, and other forms of irritations [9]. It is used for the treatment of dry and damaged skin, although it can be used for all skin types. Today, the traditional ointments with ethanolic extract of dried pot marigold flower are still used [10]. The intensive studies have focused on the development of new carriers for the topical application of pot marigold flower extract [11]. The emulsions represent the most suitable carrier so they are commonly used for the preparation of topical formulations with this extract [12, 13].

The topical formulations need to be of high quality, safe, effective, stable, and satisfy the criteria of contemporary trends. For these reasons, the use of quality, nat-
ural, and biodegradable raw materials is recommended. It is necessary to conduct characterization and evaluation of physical and chemical stability of topical formulations with herbal extracts [14, 15]. The tests of physical and chemical stability of the formulation have become important parameters that should be taken into account during the preparation, storage, and use of the product [7]. The aim of this study was to improve the properties of water-in-oil (w/o) emulsion (hydrophobic formulation), prepared using the hot/hot emulsification process with an oil phase consisted of Vaseline, lanolin, and almond oil, by the incorporation of pot marigold flower extract in different concentrations. The formulations were subjected to physical and chemical stability tests to determine the storage life and shelf-life of the products. The functional stability of the formulations was also monitored to evaluate the effect of the formulation composition on the activity of pot marigold flower extract.

**Experimental**

**Materials**

In this study, 96% (v/v) ethanol (Zorka Pharma, Sabac, Serbia), 2,2-diphenyl-1-picrylhydrazyl (DPPH), butylhydroxytoluene (BHT) (Sigma Chemical, St. Louis, Missouri, USA), Folin-Ciocalteu reagent, and gallic acid (97%) (Merck, Darmstadt, Germany) were used. The used excipients for the preparation of the w/o emulsion were: white petrolatum (white Vaseline, Vaselinum album) (Tehno-chem, Serbia), lanolin (Adeps lanae), borax, and boric acid (Sigma Aldrich, Serbia), almond water (Aqua Amygdalae Amarae), and almond oil (Oleum Amygdalae Amarae) (Sigma Chemical, St. Louis, Missouri, USA). All other chemicals were pro analysis grade.

**Plant material**

Pot marigold (Calendula officinalis L.) flowers were purchased from Dr. Josip Pancic (Belgrade, Serbia). The moisture content of 14.36% (w/w) was determined by drying plant material at 105 °C to constant weight. The dried pot marigold flowers were ground using an electric mill.

**Preparation of the extract**

Pot marigold flower extract was prepared by ultrasound-assisted extraction according to the procedure described by Zerajic et al. [16]. The ground plant material (5.0 g) was extracted with 40% (v/v) ethanol (100 mL g⁻¹) at 65 °C for 30 min. After the extraction, the solids were separated from the liquid phase by vacuum filtration on a Büchner funnel. The dry matter content of the extract was determined by drying an aliquot of 3 mL in a laboratory oven at 105 °C. Residual liquid extract was stored at 4 °C in the fridge for further analyses.

**Preparation of topical formulations**

The topical formulations (w/o emulsion) with 1% and 2% of pot marigold flower extract (F1 and F2, respectively) and cream base were prepared using the hot/hot emulsification process. The oil phase, which consisted of vaseline and lanolin, was heated in an enameled pan using a water bath at 50 ± 1 °C. Almond water was heated to the same temperature, to which different concentrations of ethanol extract were added. After heating, the aqueous phase was gradually added to the oil phase with a stirring speed of 800 rpm min⁻¹ for 5 min. The homogenization was continued at a stirring speed of 500 rpm min⁻¹ to below 40 °C when almond oil was added to the mixture. Stirring was continued until the emulsion was cooled to 22 °C (room temperature). The prepared samples were stored in a polypropylene plastic container. The cream base was prepared in the same way but without the addition of the pot marigold flower extract. The composition of topical formulations with and without pot marigold flower extract is given in Table 1.

<table>
<thead>
<tr>
<th>Ingredients</th>
<th>Cream base</th>
<th>F1</th>
<th>F2</th>
</tr>
</thead>
<tbody>
<tr>
<td>White Vaseline</td>
<td>30 g</td>
<td>30 g</td>
<td>30 g</td>
</tr>
<tr>
<td>Lanolin</td>
<td>30 g</td>
<td>30 g</td>
<td>30 g</td>
</tr>
<tr>
<td>Almond oil</td>
<td>5 g</td>
<td>4 g</td>
<td>3 g</td>
</tr>
<tr>
<td>Boric acid, 3%</td>
<td>25 g</td>
<td>25 g</td>
<td>25 g</td>
</tr>
<tr>
<td>Almond water</td>
<td>10 g</td>
<td>10 g</td>
<td>10 g</td>
</tr>
<tr>
<td>Pot marigold flower extract</td>
<td>-</td>
<td>1 g</td>
<td>2 g</td>
</tr>
</tbody>
</table>

**Methods for the evaluation of the antioxidant capacity of the pot marigold flower extract**

Folin–Ciocalteu reagent assay

Folin–Ciocalteu reagent assay (FCR) was used to estimate the reducing capacity of the ethanolic extract of pot marigold flower [17]. Gallic acid was used as a standard for the construction of the calibration curve to express the reducing capacity (grams gallic acid equivalent per 100 g of dry weight, g GAE 100 g⁻¹ d.w.). The procedure is based on the tenfold dilution of Folin–Ciocalteu reagent and its addition (1 mL) into 0.1 mL of the extract. After a few minutes, 1 mL of 7% (w/v) sodium carbonate was added to the sample. The sample was stored in a dark place for 90 min. The absorbance was measured at 765 nm on a Varian Cary 100 UV–Vis spectrophotometer (Mulgrave, Victoria, Australia).

**DPPH assay**

The antioxidant activity of the extract was determined using the DPPH assay [17]. The solution of DPPH radicals (3×10⁻⁴ mol L⁻¹) was prepared and 1 mL in 2.5 mL of extract solution was added. Instead of the extract solution, the negative control solution contained the equivalent amount of 96% (v/v) ethanol. The blank solution consisted of 1 mL of ethanol and 2.5 mL of the extract. The scanning of the samples was carried out at 517 nm after 30 min of incubation. The percentage of DPPH radicals
The thermal degradation of antioxidants in pot marigold was described using the first-order kinetics (Equation 2).

\[ C = C_0 e^{-kt} \]  

where, \( C \) – the concentration in time \( t \), \( C_0 \) – the initial concentration of the sample, and \( k \) – the first-order rate constant.

The shelf-life (\( t_{90} \)) of the extract and formulations for the first-order was calculated according to Equation 3:

\[ t_{90} = \frac{0.105}{k} \]

Testing quality of the topical formulations
Organoleptic tests
The organoleptic properties of cream base and formulations, such as appearance, color, shine, homogeneity, and phase separation were estimated by visual observation. The applicative and sensory properties of the topical formulations (ease of smearing, absorbency, stickiness, greasiness of the film on the skin) were evaluated after their application to the skin. The tests were performed 24 h after the preparation of the formulations.

Determination of the type of emulsion by measuring electrical conductivity
The electrical conductivity was measured by directly immersing the conductometer electrode (CDM 230, Radiometer, Copenhagen, Denmark) in the samples at 22 °C (room temperature). The measuring electrode was calibrated with 0.01 mol L\(^{-1}\) KCl solution.

Determination of pH values
The pH value of the aqueous phase at 22 °C (room temperature) was measured after dissolving 5 g of the sample in 25 mL of water and heating at 60 °C for 10 min. Before starting, the instrument was calibrated with standard pH buffers of 4.0 and 7.0. A pH meter (HI 9321, Hanna Instruments, Lisbon, Portugal) was used.

Viscosity
The viscosity of the prepared formulations was determined using a viscometer (Visco basic plus, Fungilab, Hapog, New York, USA) at a rotational speed of 12 rpm min\(^{-1}\) using an SP-R5 spindle.

The chemical stability of the extract and formulations was monitored at different temperatures: 4 °C, 22 °C (room temperature), and 40 °C in daylight or darkness for 3 months. The content of antioxidants was monitored spectrophotometrically using the FCR assay by measuring the absorbance at 765 nm. After exposure of 7, 30, 60, and 90 days, the FCR was determined for each sample and expressed as milligrams GAE per milliliter. The thermal degradation of antioxidants in pot marigold flower extract was described using the first-order kinetics (Equation 2).
Antimicrobial activity of extracts and topical formulations

Antimicrobial activity of pot marigold flower extract and topical formulations was studied in vitro on Gram-positive bacteria (Staphylococcus aureus ATCC 6538, Streptococcus pneumoniae ATCC 49619), Gram-negative bacteria (Escherichia coli ATCC 8739, Proteus mirabilis ATCC 25933, Klebsiella pneumoniae ATCC 10031), and fungus (Candida albicans ATCC 10231) using a disk diffusion method. The test samples were dissolved in dimethyl sulfoxide (DMSO). The antibiotic medium no. 1 was used for bacterial growth, while Sabouraud 4% Dextrose Agar was used for fungal growth. The nutrient medium was dissolved in warm purified water and then sterilized in an autoclave at 121 °C for 15 min. The sterilized medium was then cooled to 40-45 °C. The suspension of microorganisms was inserted in the medium and then poured into Petri dishes. The paper discs (6 mm in diameter) were soaked with 30 µL of the extract (11.55 mg/mL) and the sample (0.1 g mL⁻¹). Gentamicin was used as a positive control, while DMSO was used as a negative control. The bacteria were incubated at 37 °C for 18 – 24 h, while the fungi were incubated at 25 °C for 24 – 48 h under anaerobic conditions.

Statistical analysis
All obtained data are presented as the mean value of three measurements.

Results and discussion

Characterization of pot marigold flower extract
In the literature, the total polyphenolic content was mostly described as a reaction between Folin-Ciocalteu reagent and polyphenols, although this reagent can react with various bioactive compounds [20]. According to that, the proper term is the FCR. In this case, the FCR of 3.512 g GAE 100 g⁻¹ d.w. was determined for pot marigold flower extract using the spectrophotometric method. Hernández-Rosas et al. [21] determined the FCR of 37.01 mg GAE g⁻¹ for 70% (v/v) ethanolic extract of pot marigold flower prepared by maceration with stirring for 48 h. The slight differences in the FCR can be attributed to the plant growing conditions, extraction conditions, and extraction techniques.

Antioxidant activity of pot marigold flower extract
The maximum DPPH radicals inhibition of 95% was achieved at the concentration of 0.63 mg mL⁻¹ for the extract, i.e. 80% at the concentration of 0.12 mg mL⁻¹ for the synthetic antioxidant BHT after 30 min of incubation. In the literature, the aqueous extract of pot marigold flower prepared by maceration for 24 h shows maximum antioxidant activity at the concentration of 0.9 mg mL⁻¹ after 2 min of incubation [22]. The reason for differences in antioxidant activity is probably due to the use of different incubation times, extraction technique, and solvents. The IC₅₀ value of 0.14 mg mL⁻¹ obtained for the extract was higher compared to the IC₅₀ of 0.04 µg mL⁻¹ for BHT. Even though BHT was a better antioxidant, today's trend in the cosmetics industry favors the use of natural antioxidants. Given this fact, ethanol extract can be used as a source of natural antioxidants for the preparation of topical formulations.

Antimicrobial activity of the extract
The antimicrobial activity of pot marigold flower extract (11.55 mg mL⁻¹) was analyzed against different microorganism strains. The results of antimicrobial activity are shown in Table 2. The microbial strains were resistant to the effect of DMSO, while gentamicin as a positive control only affected the growth of Gram-positive and Gram-negative bacteria. The extract showed antimicrobial activity against Gram-positive (S. aureus) and Gram-negative bacteria (E. coli, P. mirabilis, and K. pneumoniae), while its effect on Gram-positive bacteria S. pneumoniae was not observed. The best inhibitory effect of the extract was noticed against E. coli that belongs to the group of resistant bacteria. It also showed an effect against C. albicans. These results are consistent with the available results [6, 23]. The antimicrobial activity of the extract enables its use as an agent in the treatment of skin infections caused by these microbes. In this way, the development of serious disorders on the skin can be prevented.

Table 2. The results of antimicrobial activity determined using a disk diffusion method

<table>
<thead>
<tr>
<th>Microorganisms</th>
<th>extract</th>
<th>F1</th>
<th>F2</th>
<th>Cream base</th>
<th>DMSO</th>
<th>Gentamicin</th>
</tr>
</thead>
<tbody>
<tr>
<td>S. aureus</td>
<td>++</td>
<td>+</td>
<td>++</td>
<td>-</td>
<td>-</td>
<td>+++</td>
</tr>
<tr>
<td>S. pneumoniae</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>++</td>
</tr>
<tr>
<td>E. coli</td>
<td>++++</td>
<td>+++</td>
<td>++++</td>
<td>-</td>
<td>-</td>
<td>++</td>
</tr>
<tr>
<td>P. mirabilis</td>
<td>++</td>
<td>+</td>
<td>++</td>
<td>-</td>
<td>-</td>
<td>++</td>
</tr>
<tr>
<td>K. pneumoniae</td>
<td>++</td>
<td>+</td>
<td>++</td>
<td>-</td>
<td>-</td>
<td>+</td>
</tr>
<tr>
<td>C. albicans</td>
<td>+++</td>
<td>++</td>
<td>++</td>
<td>-</td>
<td>-</td>
<td>-</td>
</tr>
</tbody>
</table>

The inhibition zone: <15 mm (+), 15–16 mm (+), 17–19 mm (++), 20–22 mm (+++), >23 mm (++++). Standard deviation ± 0.5 mm
Chemical stability of the extract
The temperature is a key factor in the degradation of active substances and affects the formation of hazardous products or acceleration of this process. Therefore, the chemical stability of the pot marigold flower extract was monitored at three different temperatures: 4 °C, 22 °C, and 40 °C for 90 days. In the sample of extract stored at 4 °C, a significant change in reducing capacity was not noticed, most likely due to the inhibition of phenoloxidase activity [24]. The reduction in the FCR was about 10% for the extracts stored at 22 °C and 40 °C after 90 days. It is known that the stability of polyphenols is greater at lower temperatures [25]. The rate constant of antioxidants degradation according to Equation 2 is depicted in Table 3. The maximum value of the degradation rate constant (10.6×10^{-4} day^{-1}) was reached at 40 °C, while the lowest one (4.63×10^{-4} day^{-1}) was reached at 4 °C. The obtained results indicated that the increase in temperature leads to faster degradation of antioxidants in the extract.

Table 3. The kinetic parameters of thermal degradation of antioxidants in the extract

<table>
<thead>
<tr>
<th>Temperature (°C)</th>
<th>First-order</th>
<th>R²</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>$k_r \times 10^4$ (day$^{-1}$)</td>
<td></td>
</tr>
<tr>
<td>4</td>
<td>4.63</td>
<td>0.989</td>
</tr>
<tr>
<td>22</td>
<td>9.62</td>
<td>0.986</td>
</tr>
<tr>
<td>40</td>
<td>10.60</td>
<td>0.982</td>
</tr>
</tbody>
</table>

Having in mind that the degradation of antioxidants follows the first-order kinetics [26], the shelf-life of the extract was calculated according to Equation 3. It was 7.5 months at 4 °C, 3.6 months at 22 °C, and 3.3 months at 40 °C. The temperature had a significant effect on the stability of the extract. Bilia et al. [27] obtained similar data by studying the chemical stability of two different ethanolic extracts of pot marigold flower, prepared according to the pharmacopoeial procedure, by accelerated and long-term tests. They concluded that the solvent used for the extraction affected the stability of the extract. The shelf-life of 60% (v/v) ethanolic extract was longer than 5 months and shorter than 4 months for 40% (v/v) ethanolic extract. The chemical stability of the pot marigold flower extract was evaluated taking into account the change of FCR during storage in daylight and darkness for 90 days. It was found that daylight does not have a significant impact on the degradation of antioxidants, since the change in their content was relatively small (about 3%). The obtained results enabled the selection of excipients to formulate the topical formulation and define the storage conditions of the product.

Topical formulations with pot marigold flower extract
The choice of carrier represents the important step in the formulation of topical products with herbal extracts. It affects the penetration of bioactive compounds through the skin and the loss of their activity over time. Having in mind that the emulsions are the most suitable type of carrier, one of them was used for the preparation of topical formulations with pot marigold flower extract. The oil phase of the hydrophobic formulations (w/o emulsions) consisted of white Vaseline, lanolin, and almond oil. These ingredients commonly reduce the transepidermal water loss, *i.e.* they slow down its drying. Almond oil is rich in vitamins A, B, D, and E, minerals, and essential fatty acids [28]. It helps maintain normal skin moisture, nourishes and softens the skin, reduces skin redness, and strengthens its resistance to external influences [29]. Lanolin was used as an emulsifier, but it can be also used as an antibacterial agent [30, 31]. In addition to the ingredients of the oil phase, the formulations contained a 3% (w/v) boric acid solution and almond water. Boric acid acts as an antiseptic and is used to soothe inflammatory processes on the skin [32]. The pleasant odor of the formulations originated from almond water. The formulations were prepared without the addition of preservatives, fragrances, and colors. The developed topical formulations can be used as mild antiseptics due to the presence of pot marigold flower extract. They form a thin protective layer on the skin that protects it from adverse environmental conditions. The advantage of these formulations is the absence of zinc oxide and talc that have astringent and adsorbent effects. Akhtar et al. [13] also developed a hydrophobic formulation with ethanolic extract of pot marigold flower (3%). The oil phase consisted of paraffin oil (16%) and surfactant ABIL-EM 90 (cetyl dimethicone copolyol with HLB 5) (3.5%). Almond oil was added dropwise to improve the odor of the formulation. The formulation showed moisturizing and anti-inflammatory effects.

Quality testing of the topical formulations
Organoleptic testing
The organoleptic properties of the cream base and formulations were determined 24 h after their preparation. The topical formulations with a characteristic odor were homogeneous and of semi-solid consistency. The proper selection of the oil phase ingredients and mixing speed enabled the samples to be stable 24 h after preparation. The formulations were easily smeared on the skin, leaving a greasy film that was not sticky.

Determination of the emulsion type
The electrical conductivity of the cream base was 1 μS cm$^{-1}$, while the electrical conductivity of F1 and F2 were 3 μS cm$^{-1}$ and 4 μS cm$^{-1}$, respectively. The low electrical conductivity implied that the formulations were w/o emulsions. Also, the addition of the extract in the cream base did not significantly cause the change in the electrical conductivity.

Determination of pH values
The pH value of the cream base was decreased from
5.86 to 5.81 and 5.78 for F1 and F2, respectively. This behavior was in accordance with the data reported by Akhtar et al. [13]. Since the pH values of the skin were in the allowable range of 3.5 – 8.0, the prepared formulations can be considered safe for use.

Viscosity
The viscosity of the formulations was important during the determination of the formulation consistency and evaluation of the product stability. The viscosity of the cream base, F1, and F2 was 39.5, 38.3, and 37.2 Pa s, respectively. Based on these values, it can be concluded that the formulations were of semi-solid consistency. The addition of the extract to the cream base resulted in a decrease in viscosity [33]. This is most likely caused by the internal layering in the formulation. Anchisi et al. [34] showed that the viscosity can be reduced by about 20% after adding the plant extract to the cream base.

Stability study of the formulations
The change in the FCR was monitored during the analysis of the chemical stability of the formulations. In the physical stability study, the appearance (organoleptic) and parameters characterizing the pharmaceutical-technological properties of the topical formulations (phase separation, pH value, and electrical conductivity) were also analyzed. The functional stability of the formulations was evaluated due to the change in antioxidant activity during storage. These studies were key to define the storage conditions and shelf-life of the products.

Chemical stability of the topical formulations
The change in the FCR was a key parameter for the estimation of the chemical stability of the extract after its incorporation into the topical formulations. The FCR was reduced by less than 5% after 3 months of storage, indicating the satisfactory stability of the extract in the formulations. The changes in antioxidants content in the formulations corresponded to the first-order reaction. The reaction rate constants at 22 °C for F1 and F2 were 4.846×10^{-4} day^{-1} and 4.640×10^{-4} day^{-1}, respectively. The shelf-life of developed F1 and F2, calculated according to Equation 3, was about 8 months at 22 °C (room temperature). Since the shelf-life of the extract was about 3 months, it is obvious that the improvement in the chemical stability was achieved after its incorporation into the formulations. The satisfactory stability of the extract in the formulations was the result of the proper selection of their ingredients, especially the ingredients of the oil phase.

Physical stability of the topical formulations
Centrifugation test
The prepared formulations and cream base showed satisfactory physical stability because under the influence of mechanical stress there was no phase separation.

Accelerated stability test
The pH values of cream base and formulations stored at different temperatures for 30 days are depicted in Table 4. The pH values of the cream base stored at 4 °C were increased until day 14 and then decreased. The pH values of the sample stored at 22 °C were increased with slight deviations. At 40 °C, the pH values were gradually increased for 7 days and then decreased continuously until day 30 with some deviations. At the end of this analysis, the pH values of the cream base at 4 °C, 22 °C, and 40 °C were 5.89, 6.33, and 5.54, respectively.

<table>
<thead>
<tr>
<th>Time (day)</th>
<th>4 °C</th>
<th>22 °C</th>
<th>40 °C</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>B</td>
<td>F1</td>
<td>F2</td>
</tr>
<tr>
<td>0</td>
<td>5.86</td>
<td>5.81</td>
<td>5.78</td>
</tr>
<tr>
<td>1</td>
<td>5.25</td>
<td>6.02</td>
<td>5.89</td>
</tr>
<tr>
<td>7</td>
<td>6.32</td>
<td>6.08</td>
<td>5.91</td>
</tr>
<tr>
<td>15</td>
<td>6.47</td>
<td>5.88</td>
<td>5.90</td>
</tr>
<tr>
<td>30</td>
<td>5.88</td>
<td>5.91</td>
<td>5.93</td>
</tr>
</tbody>
</table>

The pH values of F1 and F2 stored at 22 °C and 40 °C gradually increased until day 7, after which it began to decline over time with slight deviations. At 4 °C, the pH values of F1 and F2 gradually increased. At the end of day 30, the pH values of F1 were 5.91, 5.69, and 5.63 at 4 °C, 22 °C, and 40 °C, respectively, while the pH values of F2 were 5.93, 5.66, and 5.62, respectively. The decrease in pH values of the formulations at 22 °C and 40 °C may be due to the production of any acidic metabolite or the decomposition of any ingredient. The pH values of all samples after 30 days of storage at different temperatures remained in the allowed pH range of the skin. Slight changes in the pH values of the base and formulations as functions of time and temperature indicated that the tested samples were stable.

Long-term stability of the topical formulation
The values of pH and electrical conductivity of the cream base and formulations stored at 22 °C (room temperature) for 7, 30, 60, and 90 days are presented in Table 5.

The pH values were in the range 6.26 - 6.36 for the cream base, 5.72 - 5.89 for F1, and 5.66 - 5.84 for F2. The pH values for all samples after 90 days were in the allowed pH range of the skin. In contrast to the pH values, there was an increase in electrical conductivity over time. The prepared formulations remained w/o emulsion since their electrical conductivities were lower than 50 µS cm^{-1}. After 90 days of storage, there were no visible changes in appearance, color, and spreadability in the topical formulations. No significant changes in pH, electrical conductivity, and organoleptic properties were observed, indicating that the formulations were stable at 22 °C (room temperature) over time.
Antioxidant activity of the extract and functional stability of re-extracted antioxidants from topical formulations

The IC$_{50}$ value (0.14 mg mL$^{-1}$) of the extract did not differ significantly from the IC$_{50}$ value (0.15 μg mL$^{-1}$) of the extract re-extracted from the formulations. The excipients used to prepare the formulations did not alter the antioxidant activity of the extract. The functional stability of the extract, F1, and F2 stored at different temperatures for 3 months was carried out by monitoring the change in the antioxidant activity. There were no significant changes in the antioxidant activity of the pot marigold extract at 4 °C, 22 °C, and 40 °C (Figure 1a). Some deviations can be attributed to an error during the determination of antioxidant activity (coefficient of variation = 5%). The slight change in the antioxidant activity of the extract was the result of the significant stability of antioxidants. In the F1 and F2, the antioxidant activity of the extract was not significantly changed at 4 °C and 22 °C (Figure 1b and c). The loss of the activity of the extracts at 40 °C was about 8% in the F1 and F2 (Figure 1b and c). The reduction of antioxidant activity of the extract in the topical formulations was probably due to the degradation of antioxidants.

Microbiological safety of the topical formulations

The formulations were microbiologically safe because the total number of aerobic bacteria, yeast, and mold spores in 1 g of the sample was in accordance with the regulations [19]. The presence of pathogenic bacteria was not confirmed in the samples.

Antimicrobial activity of the topical formulations

The results of antimicrobial activity of cream base and formulations (0.1 g mL$^{-1}$) are depicted in Table 2. The cream base did not show inhibitory activity against the used strains of microorganisms. The developed formulations showed antimicrobial activity against Gram-positive (S. aureus) and Gram-negative bacteria (E. coli, P. mirabilis, and K. pneumoniae). The samples did not affect S. pneumoniae, while their activity was significant against E. coli. Pazhohideh et al. [35] showed that the topical formulation with pot marigold extract is effective in the treatment of bacterial vaginosis in women of reproductive age, without any side effects. Saffari et al. [36] confirmed that the Calendula vaginal cream is effective in the treatment of vaginal candidiasis. Compared to the synthetic drug clotrimazole, the formulation has a greater long-term effect.

Conclusion

The ethanolic extract of pot marigold flower with the FCR of 3.512 g GAE 100 g$^{-1}$ d.w. and expressed antioxidant (IC$_{50}$ value of 0.14 mg mL$^{-1}$) and antimicrobial activities (E. coli, C. albicans, S. aureus, P. mirabilis, K. pneumoniae) was used to improve the properties of hydrophobic formulations (w/o emulsion). The results of chemical stability indicated that the extract was more stable at lower temperatures (4 °C and 22 °C) compared to higher temperatures (40 °C). The prepared uncategorized formulations with the addition of 1% and 2% of the extract were stable during storage at different temperatures probably due to the proper selection of the formulation ingredients. They were of semi-solid consistency, homogeneous, free of synthetic odors, colors, and preservatives. The stability studies indicated that the formulations should be stored in a cool place or at controlled room temperature, avoiding prolonged exposure to temperatures above 30 °C. The microbiological analysis confirmed that the cream base and uncategorized formulations were microbiologically safe. Taking into con-

---

Table 5. The change in pH value and electrical conductivity of cream base and formulations stored at room temperature (22 °C)

<table>
<thead>
<tr>
<th>Sample</th>
<th>7 day</th>
<th>30 day</th>
<th>60 day</th>
<th>90 day</th>
<th>7 day</th>
<th>30 day</th>
<th>60 day</th>
<th>90 day</th>
</tr>
</thead>
<tbody>
<tr>
<td>B</td>
<td>6.26</td>
<td>6.30</td>
<td>6.34</td>
<td>6.36</td>
<td>1</td>
<td>2</td>
<td>3</td>
<td>3</td>
</tr>
<tr>
<td>F1</td>
<td>5.89</td>
<td>5.72</td>
<td>5.81</td>
<td>5.88</td>
<td>3</td>
<td>4</td>
<td>3</td>
<td>4</td>
</tr>
<tr>
<td>F2</td>
<td>5.84</td>
<td>5.66</td>
<td>5.74</td>
<td>5.79</td>
<td>4</td>
<td>3</td>
<td>4</td>
<td>5</td>
</tr>
</tbody>
</table>

Figure 1. The change in the antioxidant activity over time of: a) ethanolic extract, b) F1, and c) F2 at different temperatures.
sideration these findings, it can be concluded that both uncategorized formulations (F1 and F2) are suitable for topical application in the treatment of various skin infections.

Acknowledgments

The Republic of Serbia - Ministry of Education, Science and Technological Development, Program for financing scientific research work, number 451-03-9/2021-14/200133.

References


[24] L. I. Wei, Y. Zhang, The antioxidant effect of Pyracantha fortuneana polyphenol in vitro, Science and Technology of...

Izvod

**FIZIČKA, HEMIJSKA I ANTIOKSIDATIVNA STUDIJA STABILNOSTI TOPIKALNE FORMULACIJE NA BAZI EKSTRAKTA CVETA NEVENA (Calendula officinalis L.)**

Ivan M. Savić*, Ivana M. Savić Gajić

(Tehnički fakultet u Leskovcu, Univerzitet u Nišu, Bulevar oslobođenja 124, 16000 Leskovac, Srbija)

Ekstrakt cveta nevena (Calendula officinalis L.) ima farmakoška svojstva zbog prisustva različitih bioaktivnih jedinjenja. Poznato je da ekstrakt ima antioksidativna, antiinflamatorna, antimikrobijska, antifungalna, antivirusa, antimitagen, antidermatitinsa svojstva, itd. Cilj ovog rada bio je poboljšanje kvaliteta izabrane topikalne formulacije dodatkom etanolnog ekstrakta cveta nevena, kao i praćenje njene stabilnosti. Topikalna formulacija bila je emulzija voda-u-ulju pripremila određena DPPH tesc. Pripremljene nekategorisane topikalne formulacije koje sadrže 1% i 2% (m/m) ekstrakta cveta nevena bile su stabilne na različitim temperaturama tokom čuvanja. Nekategorisane formulacije su pokazivale antioksidativnu aktivnost, ali se aktivnost ekstrakta u formulacijama smanjivala sa povećanjem temperature čuvanja. Ekstrakt cveta nevena i razvijene nekategorisane formulacije su pokazivale antifungalnu aktivnost na Gram-pozitivnim (Staphylococcus aureus) i Gram-negativnim bakterijama (Escherichia coli, Proteus mirabilis, Klebsiella pneumoniae), kao i na Candida albicans. Nekategorisane formulacije sa ovom aktivnošću se mogu koristiti kod tretmana kožnih infekcija.